





SSR 5<sup>th</sup> CYCLE 2023

**2.6.1** Programme Outcomes (POs) and Course Outcomes (Cos)

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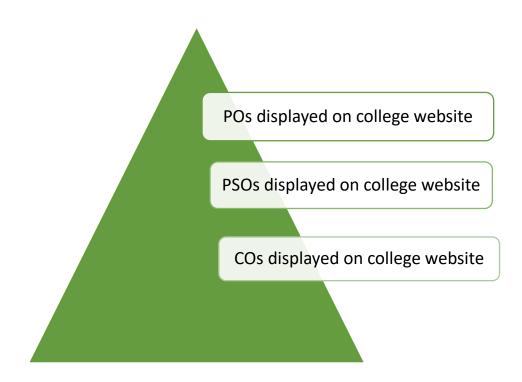
Programme Outcomes (POs) and Course Outcomes (COs) for all programmes offered by the institution are stated and displayed on website and attainment of POs and COs are evaluated – Communication of POs and COs

#### Communication of POs, PSOs and COs

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SI No	Contents
1	POs, PSOs, COs displayed on college website
2	OBE committee formed at College and Department level
3	POs-PSOs-COs display board in department - Sample
4	POs, PSOs, COs displayed on department notice boards - Sample
5	POs, PSOs, COs displayed in classroom - Sample
6	Department OBE Handbook - Sample
7	Familiarization of POs, PSOs and COs to students – Offline and Online Mode
8	First year induction program on PO-PSO-CO
9	OBE question paper stating COs - Sample
10	Workshops and training on OBE conducted for teachers

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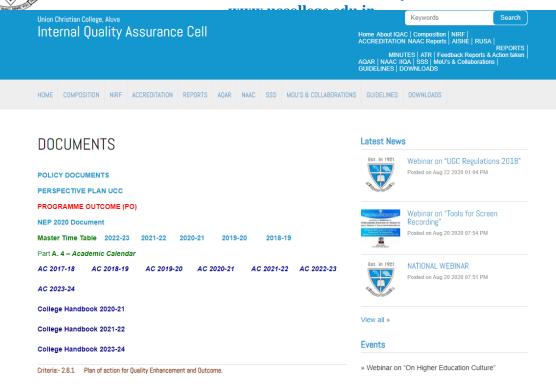
#### 1.POs, PSOs, COs displayed on college website





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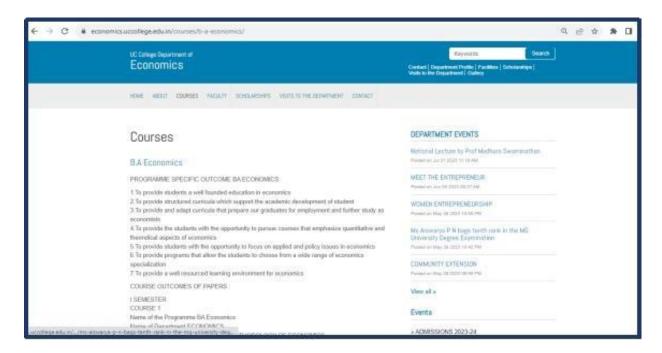
POS DISPLAYED ON COLLEGE WEBSITE

Link: https://iqac.uccollege.edu.in/documents/



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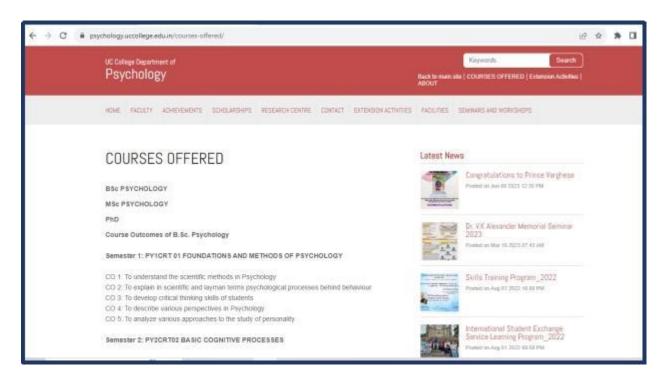
• PSOs DISPLAYED ON DEPARTMENT WEBSITE-ECONOMICS DEPARTMENT(SAMPLE)



Link: https://economics.uccollege.edu.in/courses/b-a-economics/

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• COs DISPLAYED ON DEPARTMENT WEBSITE-PSYCHOLOGY DEPARTMENT(SAMPLE)



Link: https://psychology.uccollege.edu.in/courses-offered/



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#### 2. OBE committee formed at College and Department level



**OBE** College level committee meeting

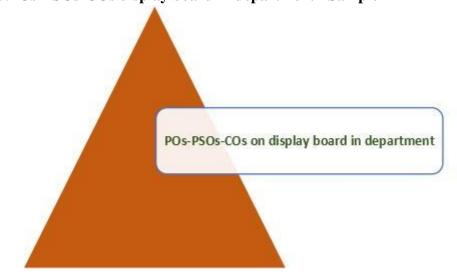


**OBE** department level meeting -sample

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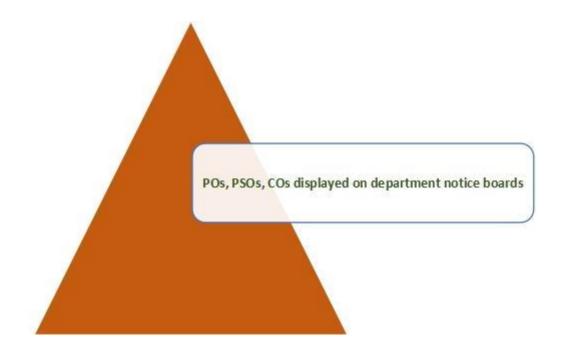
3.POs-PSOs-COs display board in department - Sample



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4. POs, PSOs, COs displayed on department notice boards - Sample



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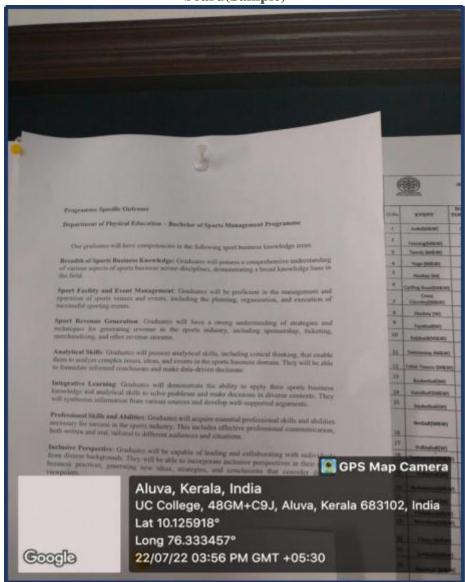
#### **PO Displayed on College**



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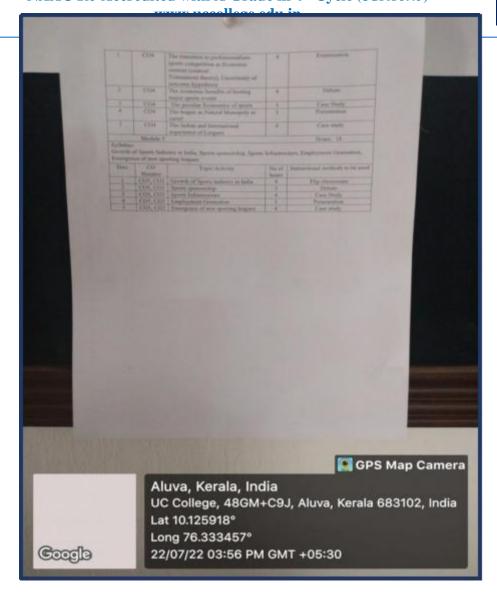
 POs, PSOs, COs displayed on Physical Education Department notice board(Sample)





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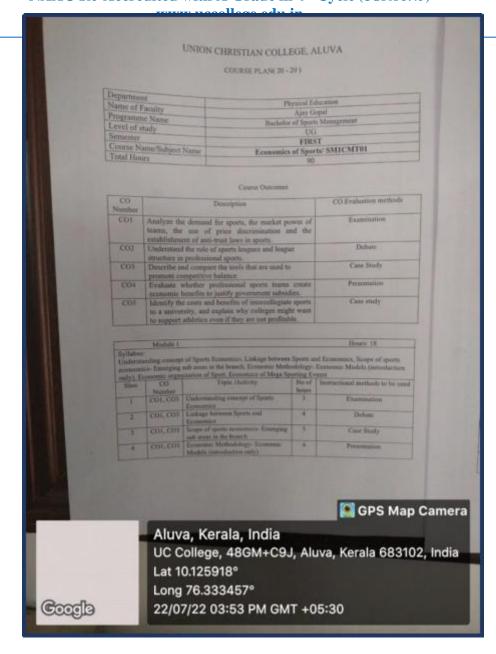
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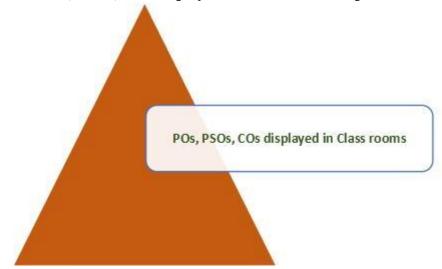
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5.POs, PSOs, COs displayed in classroom - Sample



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48GM+GGP, Aluva, Kerala 683102, India 03 Aug 2023 02:53 pm

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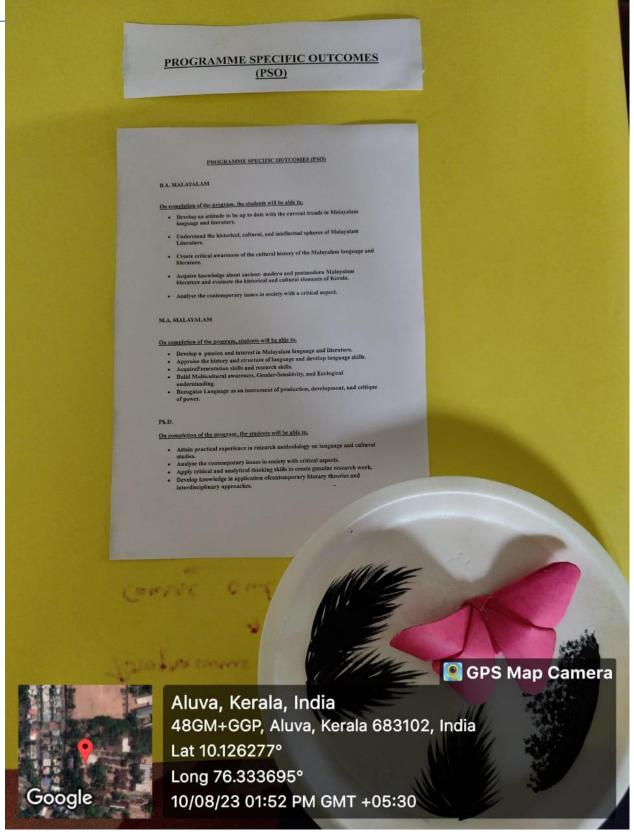
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SSR 5<sup>th</sup> CYCLE 2023

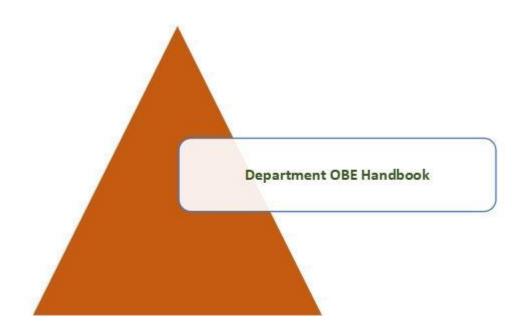




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6.Department OBE Handbook - Sample

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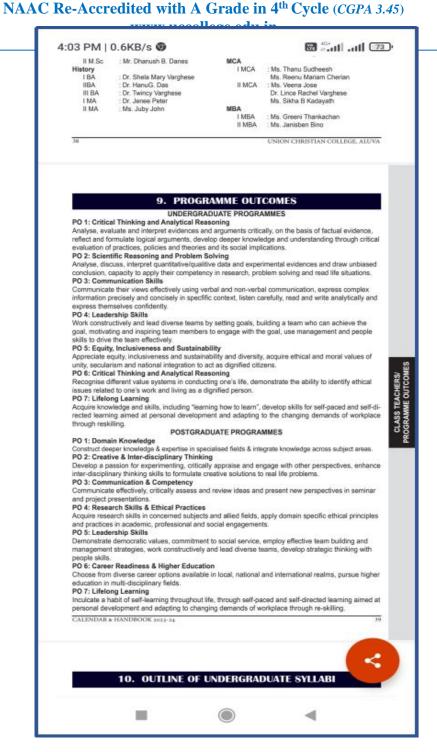


POs stated in the College Handbook sample



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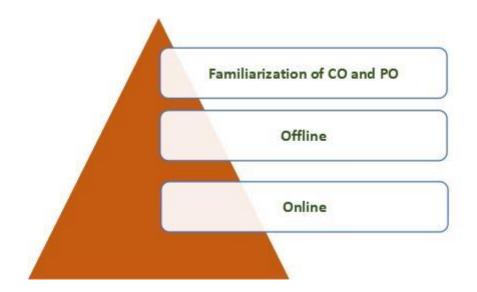
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7. Familiarization of POs, PSOs and COs to students - Offline and Online Mode

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• Offline class on CO and PO to students of B.Sc Zoology



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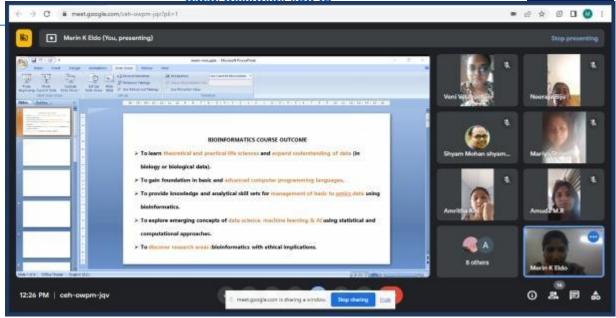


• Online: Introducing Course outcomes to first year PG Students of Biosciences

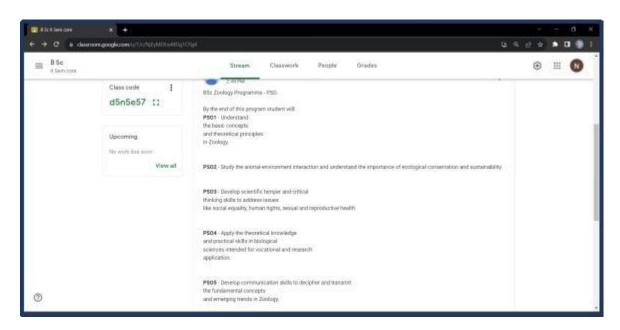


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• Resource material on PSO and CO in Google Classroom



8.First year induction program on PO-PSO-CO



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**COLLEGE LEVEL ORIENTATION** 

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For parents and Students

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**For Parents** 



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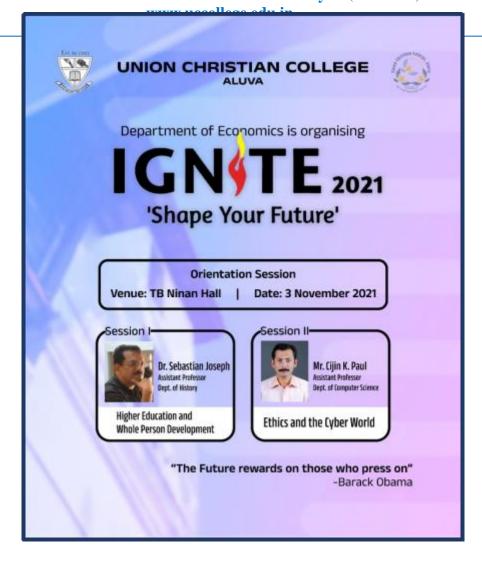
SSR 5th CYCLE 2023

**DEPARTMENT LEVEL ORIENTATION** 



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Orientation to first year students on OBE and Whole Person Development



Orientation and Inauguration for MCA freshers by Ernakulam Collector Dr. Renu Raj

9.OBE question paper stating COs - Sample



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OBE question paper stating COs -Sample

	Union Christ	tian College, A		ode No: (	606
	First Internal Exam Ma	rch 2023			
Course Code: CH2CRT02 Course Title: Theoretical a			and Inorganic Chemistry		
Facu	lty in charge: Dr. Jenish Paul & Ms.Minu Joys	Max Marks: 30	Duration: 1 1/2 Hours		
Name:		Class No./ Reg No: Name of program:		Semester: II	
CO2	se Outcome: : Summarise various models of atomic structure : Apply various theories of chemical bonding : Apply and analyse basic ideas on s-block, p-block, d-b	block and f-block elements			
SI No	Question		Bloom's Taxonomy (Revised)	CO	Marks
Sect	ion A Answer any five of the following (1 mar	k for each question) (1X5=5	marks)		
1	The radius of first orbit of hydrogen atom is 0.59 A <sup>0</sup> . hydrogen atom is	Apply	COI		
2	Define lattice energy		Remember	CO2	
3	Discuss J.J.Thomson's model of atom		Understand	COI	
4	Analyse Dalton Atomic theory		Analyse	COI	
5	Use Fajan's rule to the polarization of ions		Apply	CO2	
6	Deduce any two properties of ionic bond		Evaluate	CO2	
Sect	ion B Answer any three of the following (5 ma	arks for each question) (5X3	=15 marks)		
7	Describe blackbody radiation		Understand	COI	
8	Examine the important features of Rutherford atom model.		Analyze	COI	
9	Deduce Born-Lande equation		Evaluate	CO2	
10	Apply Einstein equation to explain the photoelectric effect		Apply	COI	
11	Examine the various factors affecting the formation of ionic bond		Analyze	CO2	
Sect	ion C Answer any one of the following (10 ma	rks for each question) (10X	1=10 marks)		
12	Using the postulates of Bohr model, find the radius of deduce the energy of electron in hydrogen atom.	Apply	CO1		
13	Illustrate the Born- Haber cycle and its application	Apply	CO2		

**Sample Internal Question Paper** 

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#### 10. Workshops and training on OBE conducted for teachers





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Webinar on OBE organized by Dept.of Chemistry

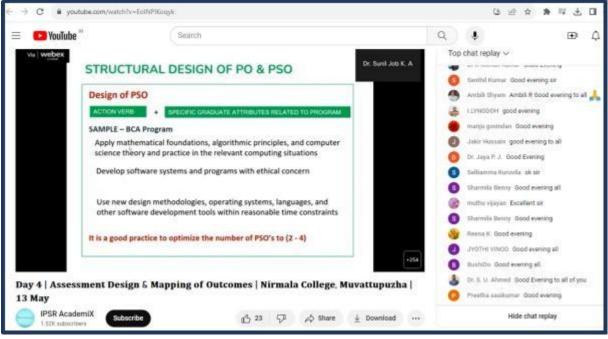


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Certificate received by Liji Lawrence for attending FDP on OBE



**Faculty Training on OBE- Sample sessions**